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# The Effect of Cation Ion Ordering in In- Cr- Sulphur Spinel.

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## Introduction

Chromium chalcogenide spinel  $MCr_2X_4$  (M=Fe, Co, Cu, Cd, X=S, Se) shows various magnetic propertie with M ions.  $CuCr_2Se_4$  and  $CdCr_2Se_4$  are known to show metallic conduction and large magneto-optical effect [1]. In addition to colossal magnetoresistane (CMR) effect, metal-insulator transition and structural phase transition appear in  $FeCr_2S_4$  [2], spin-frustration effects reveal in  $FeSc_2S_4$  and  $MnSc_2S_4$  [3]. These systems have been revisited relaxor ferroelectricity and colossal magnetocapactive effect [4]. These features were attributed to competition of isomorphic ions with the topological frustration, Jahn-Teller distortion, and geometric frustration of magnetic moment. Here, we report the magnetic properties of the  $FeCr_2S_4$  and  $FeIn_2S_4$  with special emphasis on cation ordering related to the quadrupole interactions.

# **Experiments**

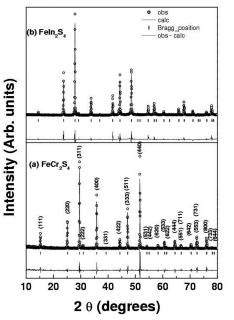
Syntheses of the  ${\rm FeCr_2S_4}$  and  ${\rm FeIn_2S_4}$  were accomplished by the solid state reaction of the highpurity elements Fe, Cr, In and S in an evacuated quartz tube. The crystal structure of the samples was examined using x-ray diffractometer (XRD) with  ${\rm Cu}$ - $K_{\alpha}$  radiation and analyzed by Rietveld refinement. The Mössbauer spectra were recorded using a conventional spectrometer of electromechanical type with a  $^{57}{\rm Co}$  source in a rhodium matrix.

# Results and discussions

Fig.1 shows the x-ray diffraction refinements for the  $FeCr_2S_4$  and  $FeIn_2S_4$  samples at room temperature, respectively. The spectra shown in Fig. 1 demonstrate the absence of any impurity phases. The determined crystal symmetry of samples is a cubic spinel structure Fd3m.  $FeIn_2S_4$  is an inverse spinel, with In atoms occupying both tetrahedral (A) and octahedral (B) sites. On the other hand,  $FeCr_2S_4$  has a normal spinel with Fe atoms occupying A site and Cr atoms occupying B site. The determined lattice constant  $a_0$  for  $FeCr_2S_4$  and  $FeIn_2S_4$  were  $a_0$ =10.011 and  $a_0$ =10.616 Å, respectively, since result might be the larger ionic radius for In ions than for Cr ions.

In order to study microscopic interaction mechanism, the Mössbauer spectra of sulphur spinel FeCr<sub>2</sub>S<sub>4</sub> and FeIn<sub>2</sub>S<sub>4</sub> have been studied. Fig. 2 shows Mössbauer spectra for the FeCr<sub>2</sub>S<sub>4</sub> and FeIn<sub>2</sub>S<sub>4</sub> at 4.2 K and room temperature, respectively. The Néel temperatures were found to be 175 and 15 K for the FeCr<sub>2</sub>S<sub>4</sub> and FeIn<sub>2</sub>S<sub>4</sub>, respectively, by Mössbauer spectroscopy. It can be understood as the strength of inter-sublattice exchange interaction Fe<sup>2+</sup>(A)-S<sup>2-</sup>-Cr<sup>3+</sup>(B) is stronger than that of the intra-sublattice exchange interaction Fe<sup>2+</sup>(B)-S<sup>2-</sup>-Fe<sup>2+</sup>(B). The large asymmetrical line broadening of Mössbauer absorption lines is shown for the samples at 4.2 K. We note that the FeCr<sub>2</sub>S<sub>4</sub> shows a single line resonance spectrum with an isomer shift of 0.72 mm/s at room temperature, while FeIn<sub>2</sub>S<sub>4</sub> at room temperature has an isomer shift of 0.74 mm/s and a electric quadrupole splitting ( $\Delta E_Q$ ) of 3.22 mm/s. The charge state of Fe ions is ferrous (Fe<sup>2+</sup>) as characterized by isomer shift ( $\delta$ ) for the samples. We interpret that the presence of the  $\Delta E_Q$  is attributed to the trigonal field at the octahedral site, according to Fe<sup>2+</sup> ions enter to octahedral B site.

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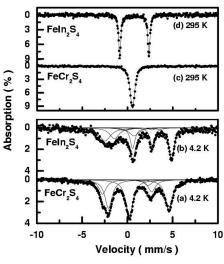


Fig. 2. Mössbauer spectra of FeCr<sub>2</sub>S<sub>4</sub> and FeIn<sub>2</sub>S<sub>4</sub> at 4.2 K and room temperature.

Fig. 1. Refined x-ray diffraction patterns of the (a) FeCr<sub>2</sub>S<sub>4</sub> and (b) FeIn<sub>2</sub>S<sub>4</sub> at room temperature.