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ABSTRACTS

**HR-04. The effect of manganese ions in  $\text{MnCr}_{1.98}^{57}\text{Fe}_{0.02}\text{O}_4$  by Mössbauer spectroscopy.** K. Choi<sup>1</sup>, S. Kim<sup>1</sup> and C. Kim<sup>1</sup>. *Physics, Kookmin Univ., Seoul, South Korea*

The main scope of the present study is to examine the weak geometrical frustration and multiferroic effect in  $\text{M Cr}_2\text{O}_4$  ( $\text{M} = \text{Co}, \text{Mn}$ ) [1, 2].  $\text{Co}^{2+}$  and  $\text{Mn}^{2+}$  ions occupy the A site,  $\text{Mn}^{2+}$  ( $3d^5$ ) exhibits a half-filled shell with a total spin  $S = 5/2$  and zero orbital moment and  $\text{Co}^{2+}$  ( $3d^7$ ) has  $S = 3/2$  and small spin-orbit effects [3]. Geometrical frustration was affected among the B sites forming the pyrochlore lattice relation with magnetic ions of A sites. We have substituted a small amount of Fe ions for Cr sites and investigated the magnetic behavior of Fe ions, on atomic scale, using Mössbauer measurement. Polycrystalline  $\text{MnCr}_{1.98}^{57}\text{Fe}_{0.02}\text{O}_4$  compound was prepared by wet-chemical process. The crystal structure was found to be single-phase cubic spinel with space group of  $\text{Fd}\bar{3}m$ . The lattice constant  $a_0$  and the internal structural parameter ( $x$ ) of the oxygen were determined to be 8.444 Å and 0.263, respectively. Mössbauer spectra of  $\text{Mn}_{1.98}^{57}\text{Fe}_{0.02}\text{O}_4$  were taken from 4.2 to 295 K. The absorption spectra at 4.2 K show that the well developed two sextets are superposed with small difference of hyperfine field ( $H_{\text{hf}1} = 485$  and  $H_{\text{hf}2} = 475$  kOe). Isomer shift values ( $\delta$ ) of doublets are found to be 0.42 and 0.43 mm/s relative to the Fe metal, respectively, which are consistent with the high spin  $\text{Fe}^{3+}$  charge state. Above the Néel temperature ( $T_N = 45$  K) paramagnetic doublet is observed. We note that even though the magnetic hyperfine fields on B-site are similar a noticeable difference of the Néel temperatures is shown between two samples ( $\text{M}_{1.98}^{57}\text{Fe}_{0.02}\text{O}_4$  ( $\text{M} = \text{Co}, \text{Mn}$ )). It is interpreted in terms of A-B super-exchange interaction with related ionic bond lengths.

[1]. K. Tomiyasu, et al., Phys. Rev. B., 74, 024413 (2004). [2]. Y. Yamasaki, et al., Phys. Rev. Lett. 96, 207204 (2006). [3]. N. Tristan, et al., Phys. Rev. B., 72, 174404 (2005).