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Ba<sub>2</sub>Co<sub>2</sub>Fe<sub>12</sub>O<sub>22</sub>. I. Lee<sup>1</sup>, H. Cho<sup>1</sup> and C. Kim<sup>1</sup>I. Department of Physics, Kookmin University, Seoul, Korea, Republic of The cobalt substituted Y-type barium ferrite Ba<sub>2</sub>Co<sub>2</sub>Fe<sub>12</sub>O<sub>22</sub> (Co<sub>2</sub>Y) was prepared by solid state reaction method. From the refined X-ray diffraction patterns, there are six interstitial sites for Fe and Co ions such as  $3b_{yy}$ ,  $6c_{yy}$ ,  $6c_{\text{vi}}$ ,  $18h_{\text{vi}}$ ,  $6c_{\text{iv}}$  and  $3a_{\text{vi}}$ . Also, the crystal structure was found to be a single-phase rhombohedral structure with the lattice constants  $a_0=5.8638$  Å and  $c_0$ =43.5259 Å (space group: R-3mH). The Bragg factor  $R_{\rm B}$  and  $R_{\rm F}$  were 5.83% and 3.81%, respectively. The Mössbauer spectra of  $Co_2Y$  were taken in the temperatures range 4.2 K  $\leq T \leq 715$  K. The spectra below Curie temperature ( $T_a=715$  K) were fitted by a least-squares technique with six interstitial Fe sites corresponding to the  $3b_{yy}$ ,  $6c_{yy}$ ,  $6c_{yy}$ ,  $18b_{yy}$ ,  $6c_{yy}$  and  $3a_{yy}$  subspectrum. From the analyzed Mössbauer spectrum, the site occupancy in six interstitial sites of the Co and Fe ions were calculated by the relative subspectrum absorption areas.[1] The occupation number of Co ions in the system determined to be 0.55 and 0.35 for  $18h_{yy}$  and  $6c_{yy}$  sites, respectively. This result is an experimental evidence of the site occupancy distribution in the six interstitial Fe<sup>3+</sup> sites for Ba<sub>2</sub>Co<sub>2</sub>Fe<sub>12</sub>O<sub>22</sub>.

[1] Z. W. Li, L. Guoqing, N.-L. Di, Z.-H. Cheng, and C. K. Ong, Phys. Rev.

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DW-10. Mössbauer spectra of the Co substituted Y-type Ba-ferrite