

Examination of the magnetic hyperthermia and other magnetic properties of $\text{CoFe}_2\text{O}_4@ \text{MgFe}_2\text{O}_4$ nanoparticles using external field Mössbauer spectroscopy

Jeongho Park, Hyunkyung Choi, Sam Jin Kim, and Chul Sung Kim^a
Department of Physics, Kookmin University, Seoul 02707, South Korea

(Presented 7 November 2017; received 2 October 2017; accepted 31 October 2017;
published online 21 December 2017)

$\text{CoFe}_2\text{O}_4@ \text{MgFe}_2\text{O}_4$ core/shell nanoparticles were synthesized by high temperature thermal decomposition with seed-mediated growth. The crystal structure and magnetic properties of the nanoparticles were investigated using X-ray diffractometry (XRD), vibrating sample magnetometry (VSM), and Mössbauer spectrometry. The magnetic hyperthermia properties were investigated using a MagneTherm device. Analysis of the XRD patterns showed that $\text{CoFe}_2\text{O}_4@ \text{MgFe}_2\text{O}_4$ had a cubic spinel crystal structure with space group $Fd-3m$ and a lattice constant (a_0) of 8.3686 Å. The size and morphology of the $\text{CoFe}_2\text{O}_4@ \text{MgFe}_2\text{O}_4$ nanoparticles were confirmed by HR-TEM. The VSM measurements showed that the saturation magnetization (M_S) of $\text{CoFe}_2\text{O}_4@ \text{MgFe}_2\text{O}_4$ was 77.9 emu/g. The self-heating temperature of $\text{CoFe}_2\text{O}_4@ \text{MgFe}_2\text{O}_4$ was 37.8 °C at 112 kHz and 250 Oe. The $\text{CoFe}_2\text{O}_4@ \text{MgFe}_2\text{O}_4$ core/shell nanoparticles showed the largest saturation magnetization value, while their magnetic hyperthermia properties were between those of the CoFe_2O_4 and MgFe_2O_4 nanoparticles. In order to investigate the hyperfine interactions of CoFe_2O_4 , MgFe_2O_4 , and $\text{CoFe}_2\text{O}_4@ \text{MgFe}_2\text{O}_4$, we performed Mössbauer spectrometry at various temperatures. In addition, Mössbauer spectrometry of $\text{CoFe}_2\text{O}_4@ \text{MgFe}_2\text{O}_4$ was performed at 4.2 K with applied fields of 0–4.5 T, and the results were analyzed with sextets for the tetrahedral A-site and sextets for the octahedral B-site. © 2017 Author(s). All article content, except where otherwise noted, is licensed under a Creative Commons Attribution (CC BY) license (<http://creativecommons.org/licenses/by/4.0/>). <https://doi.org/10.1063/1.5007347>