

Effect of Mg Shallow Doping on Structural and Magnetic Properties of LiFePO₄ Triphylite

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We examine the structural and magnetic properties of Mg shallow-doped LiMg_xFe_{1-x}PO₄ ($x = 0.01, 0.05, \text{ and } 0.1$) samples synthesized by the conventional solid-state reaction route. Rietveld refinement of the X-ray diffraction patterns shows the crystal structure and bond lengths of LiO₆ and Fe(Mg)O₆. The Néel temperature (T_N), spin reorientation temperature (T_S), Curie–Weiss temperature, and effective moment decreased owing to the decrease in superexchange interactions with increasing Mg²⁺ ions. The Mössbauer spectra of samples below T_N were fitted with asymmetrical eight absorption lines, whereas above T_N were fitted with symmetrical doublet. The change in the value of electric quadrupole splitting and its slop of magnetic susceptibility at T_S suggests that spin reorientation related to the orbital angular moment contributes to the magnetic properties by spin–orbit coupling. The value of isomer shift of samples at all temperatures was Fe²⁺ ion states and it decreased with increasing Mg²⁺ ion owing to decrease in charge density at the Fe nucleus due to the presence of Mg²⁺ ions on the FeO₆ site.

Index Terms—Antiferromagnetic ordering, cathode, Mössbauer spectroscopy, spin reorientation.